

Highly luminescence hybrid perovskite: Al_2O_3 composites

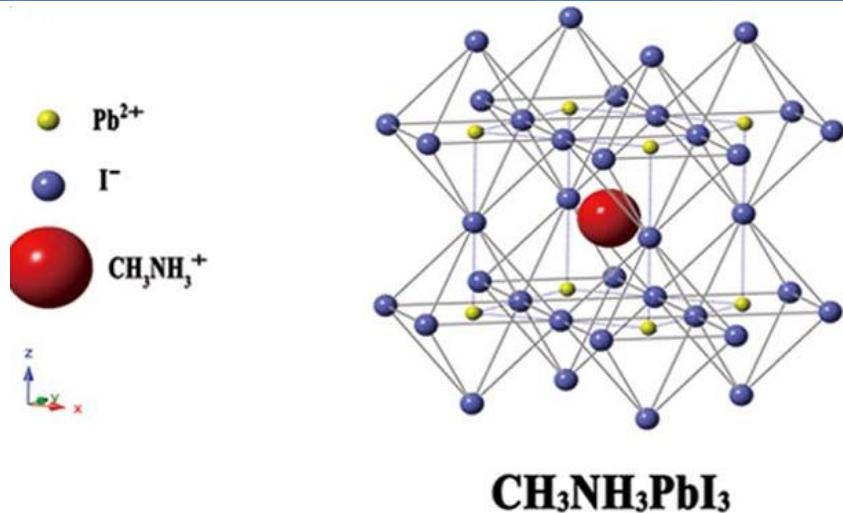
Giulia Longo

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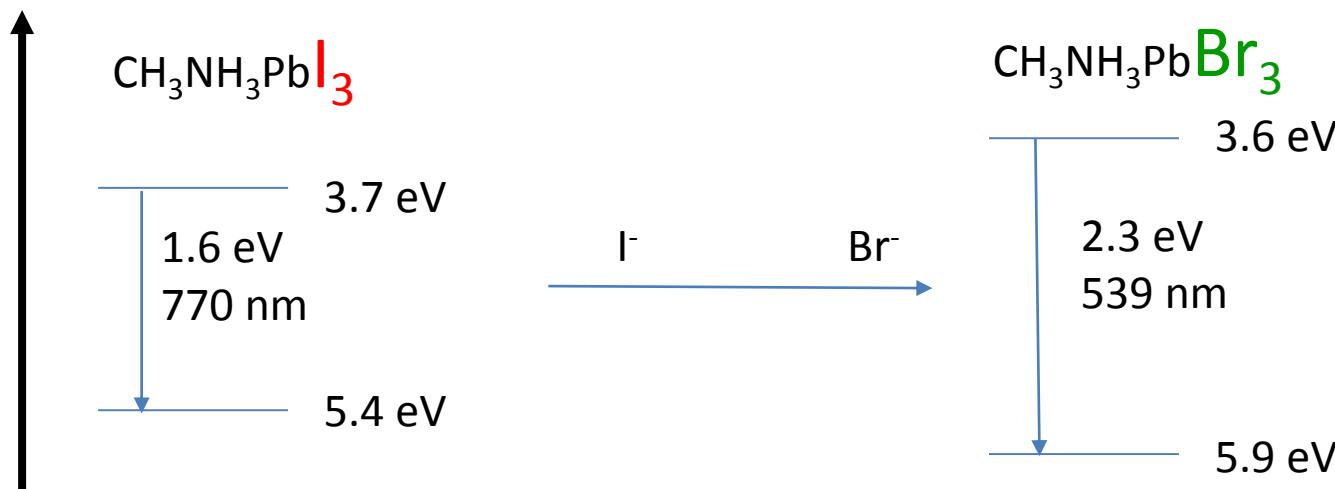
28.10.2015

Organic-inorganic cubic perovskites

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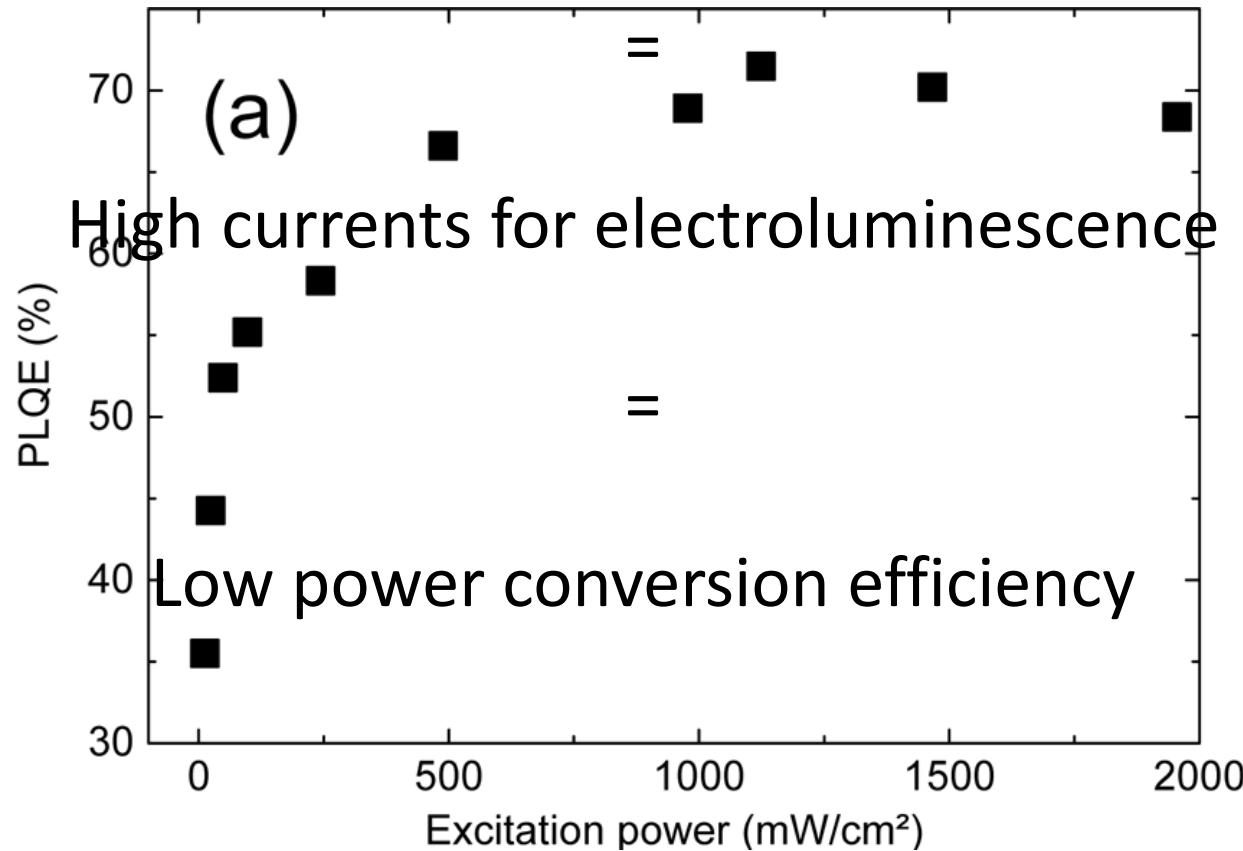
Corner sharing lead halide octahedra
 (PbX_2)
+
Methylammonium lead halide
 $(\text{CH}_3\text{NH}_3\text{X})$



Why high photoluminescence?

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High PL with high excitation intensities



Increasing the PL by size control

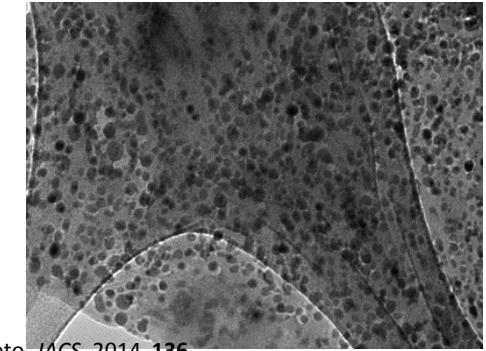
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Low photoluminescence from bulk material

Formation of nanoparticles

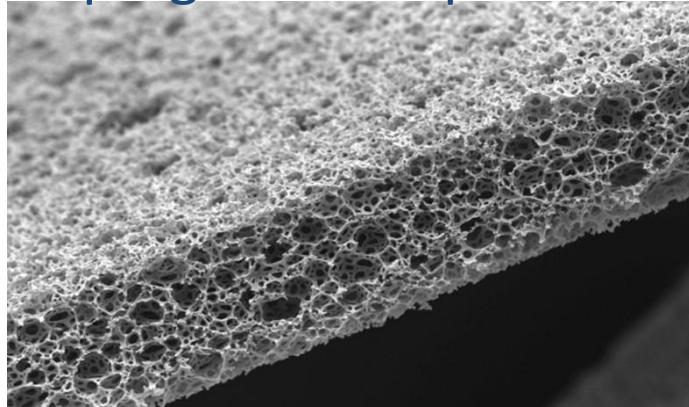


Blending with polymers



L.C. Schmidt, A. Pertegás, S. González-Carrero, O. Malinkiewicz, S. Agouram, G. Mínguez Espallargas, H.J. Bolink, R. E. Galian, J. Pérez-Prieto, *JACS*, 2014, **136**
G. Li, Z.K. Tan, D. Di, M.L. Lai, L. Jiang, J.H.W Lim, R.H. Friend, N.C. Greenham, *Nano Lett.*, 2012

Impregnation in porous scaffold

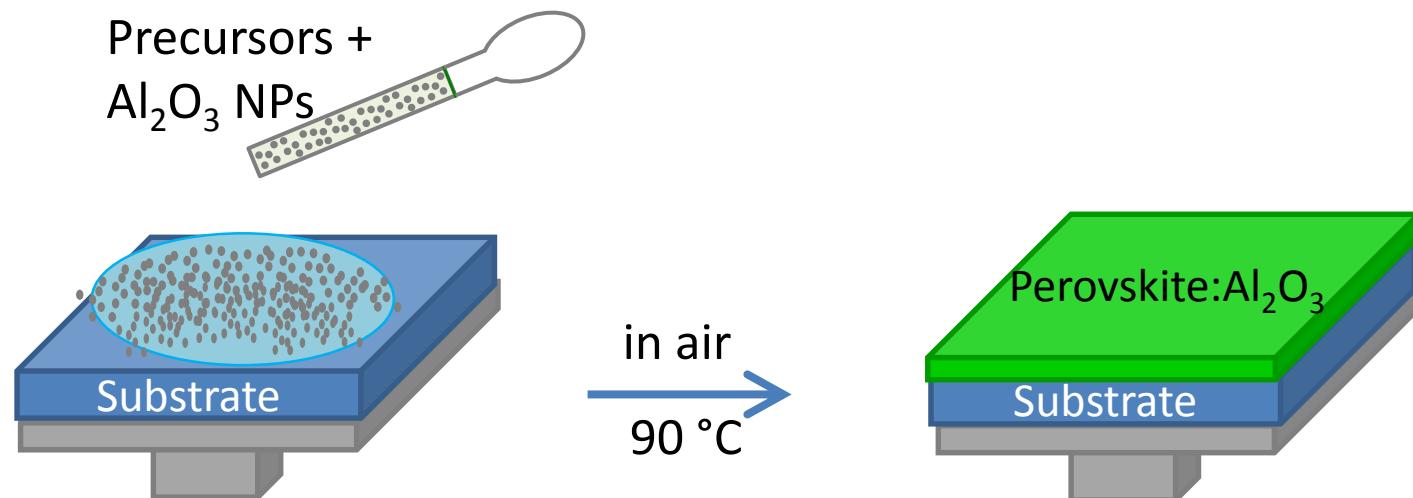


Our results

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Blending Al_2O_3 NPs with perovskite.

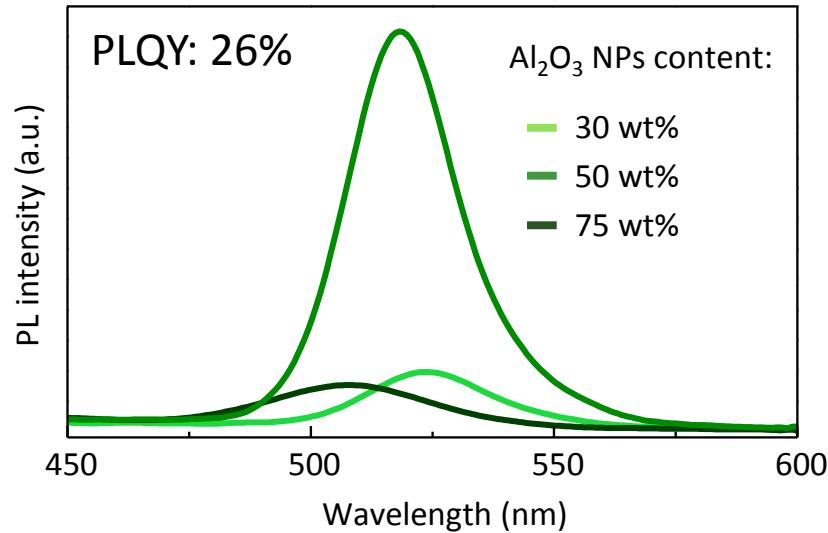
Porous scaffold formed at low temperature



Photoluminescence

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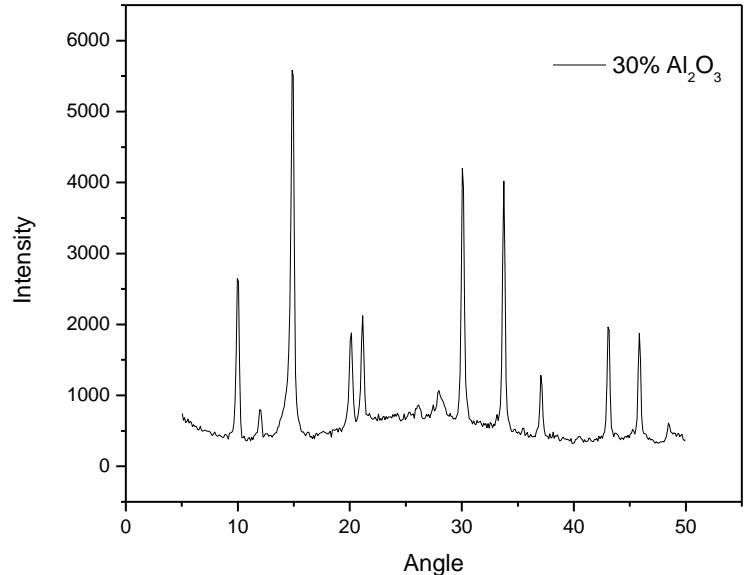
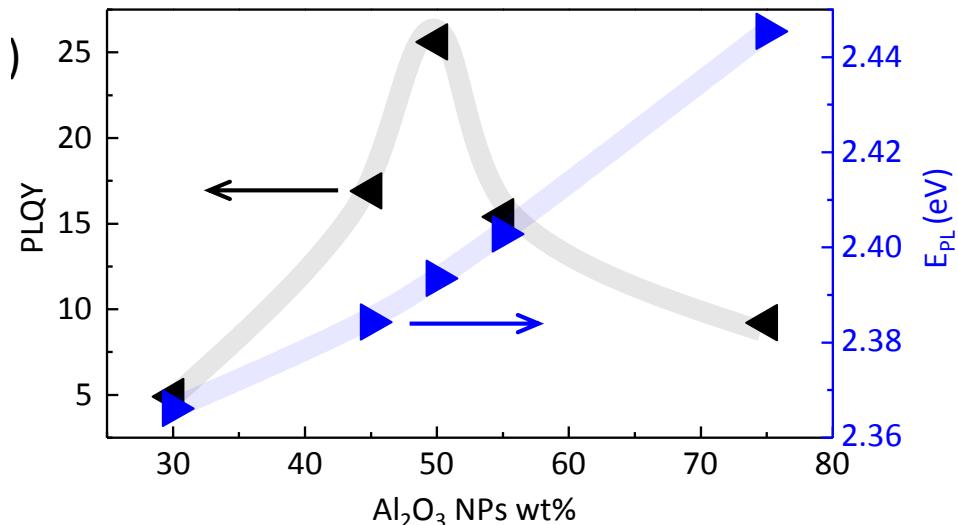
Spin coating of the blend on quartz.
Annealing at 90C for 1h



With annealing optimization of
the 50% composite we reached
39% of PLQY

Blueshift

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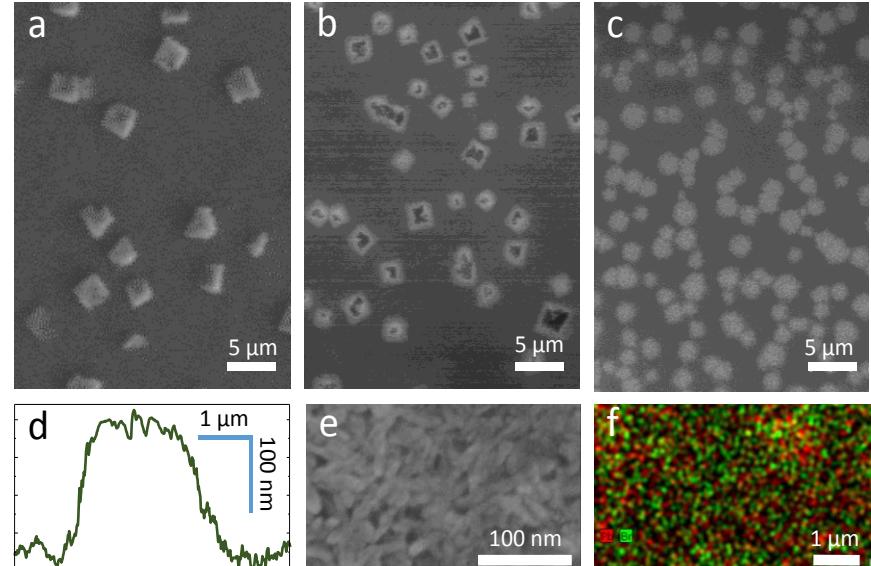
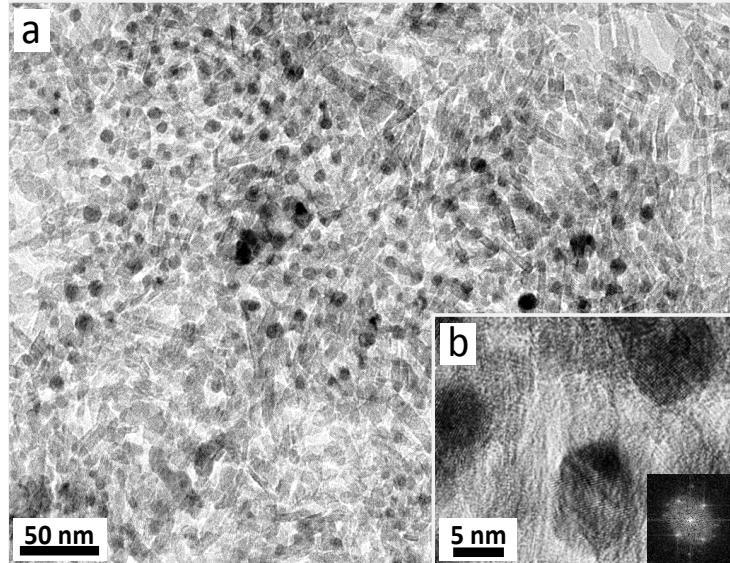
Scherrer equation: 20 nm

Theoretical bohr radii: 2.2 nm

Br-Pb-Br distortion

Morphology

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Presence of two type of perovskite crystals: nanocrystals and capping crystals

- Nanocrystal: visible by TEM, and confirmed by the energy dispersive X-Ray analysis (EDXA)
Responsible of light emission
- Capping layer: visible by TEM and AFM. Big crystals that are not responsible to the light emission.

Acknowledgments

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- EU Marie Curie LUMINET project
- University of Valencia, ICMol
- My supervisors, Henk Bolink and Michele Sessolo
- Dr. Gerlitzki for the help during the Secondment

Thank you for your attention!